Preparations of Perhydroimidazo[1,5-*a*]pyridine-3-thiones and Thiazole-5spirocyclopropan-4(5*H*)-ones from Thioureas and α , γ -Dibromobutyryl Chloride under Phase-transfer Conditions

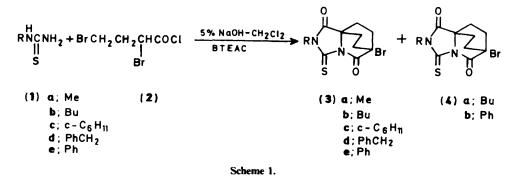
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The reaction of *N*-monosubstituted thioureas (1) with α,γ -dibromobutyryl chloride (2) was carried out in 5% NaOH–CH₂Cl₂ to afford 2-alkyl(aryl)-6-bromo-2,3,7,8-tetrahydro-3-thioxo-6,8a-ethanoimidazo[1,5-a]pyridine-1,5(6H,8aH)-diones (3) and 2-alkyl(aryl)-aminothiazole-5-spirocyclopropan-4(5H)-ones (5).

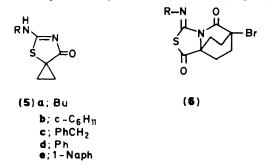
Although the formation of four-, five-, and six-membered heterocyclic compounds by the reaction of thioureas with α - and β -halogenocarboxylic acid derivatives has been well documented,¹⁻⁵ there has been no report of the synthesis of heterocycles from thioureas and α,γ -dihalogenocarboxylic acid derivatives. We now report the novel synthesis of 2-alkyl(aryl)-6-bromo-2,3,7,8-tetrahydro-3-thioxo-6,8a-ethanoimidazo[1,5-*a*]pyridine-1,5(6H,8aH)-dione (3) and 2-alkyl(aryl)amino-oxothiazole-5-spirocyclopropane-4(5H)-ones (5) by t reating N-monosubstituted thioureas (1) with α,γ -dibromobutyryl chloride (2) under phase-transfer conditions.

1 728—1 740 and 1 640—1 660 cm⁻¹. Several other isomeric structures for the product are also possible, although all except (3) and (6) are extremely strained; in view of this the carbonyl absorptions of (3) and (6) would be expected to be at lower frequency than those for the other isomers. The thioureido absorption was observed ⁶ at 1 500—1 520 cm⁻¹, a result which is supportive of structure (3) over that of (6). The ¹³C n.m.r. spectra exhibited carbonyl signals at 185.5 and 182.1 p.p.m., and a thione signal at 191.9 p.p.m.⁷ The ¹H n.m.r. and mass spectral data also supported the structural assignment of the product as (3).



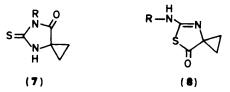
The reaction was successfully carried out by treating the chloride (2) with the thiourea (1) in 5% NaOH-CH₂Cl₂ in the presence of benzyltriethylammonium chloride at room temperature. The resulting products (3) and (5) were purified by silica-gel column chromatography (CHCl₃-benzene, 8:2). The results are shown in Table 1 and 2.

The structure of compound (3) was assigned as follows. On the basis of mass spectral data and elemental analyses, it is evident that the product was obtained by the reaction of compounds (1) with (2) in the ratio of 1:2. The i.r. spectra showed absorptions assignable to two carbonyl groups at



For compounds with $\mathbf{R} = \mathbf{Bu}$ and $\mathbf{c}-\mathbf{C}_{6}\mathbf{H}_{11}$, the debromo compounds (4) derived from compounds (5) were isolated. Structural assignments were made on the basis of a comparison of their spectral data compared with those of compounds (3). However, the path for the debromination is unclear.

In assigning a structure for product (5), the isomeric structures (7) and (8) were also considered. The i.r. spectra of product (5) showed carbonyl and C=N absorptions at 1 670—1 700 and 1 640 cm⁻¹, respectively, but thioureido absorption



was absent. This denies the possibility of the structure being (7), but can not discriminate between structures (5) and (8). Therefore, the hydrolysis of the product was carried out in 15%NaOH-EtOH under reflux to afford the disulphide (9) and amine. This result supports the assigned structure (5).

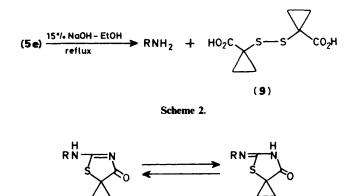
							Found (%) (Required)		
(3)	M.p. (°C)	Yield (%)	$v_{max.}(KBr) (cm^{-1})$	δ_{H} in CDCl ₃	$m/z (M^+)$	Formula	΄c	н	Ň
8	127—128	32	1 729 (C=O) 1 640 (C=O) 1 510 [NC(S)N]	1.40—1.97 (m, CH ₂ × 4, 8 H) 3.30 (s, CH ₃ N, 3 H)	302 304	$C_{10}H_{11}BrN_2O_2S$	39.90 (39.62)	3.68 (3.66)	9.08 (9.24)
b	86—87	26	1 740 (C=O) 1 660 (C=O) 1 520 [NC(S)N]	0.93—1.87 (m, C_3H_7 and $CH_2 \times 4$, 15 H), 3.85 (t, CH_2N , 2 H)	344 346	$C_{13}H_{17}BrN_2O_2S$	45.06 (45.22)	4.90 (4.96)	8.18 (8.11)
c	177—178	24	1 735 (C=O) 1 640 (C=O) 1 500 [NC(S)N]	1.03—1.94 (m, $CH_2 \times 4$ and $c-C_6H_{11}$, 19 H)	370 372	$C_{15}H_{19}BrN_2O_2S$	48.49 (48.52)	5.13 (5.16)	7.52 (7.54)
d	122—123	36	1 728 (C=O) 1 655 (C=O) 1 500 [NC(S)N]	1.50 and 1.73 (m, $CH_2 \times 4, 8 H$) 4.98 (s, CH_2N , 2 H), 7.32 (s, Ph, 5 H)	378 380	$C_{16}H_{15}BrN_2O_2S$	50.95 (50.67)	3.96 (3.99)	7.37 (7.39)
e	157—158	31	1 728 (C=O) 1 655 (C=O) 1 510 [NC(S)N]	1.47 and 1.80 (m, $CH_2 \times 4$, 8 H), 7.10-7.50 (m, Ph, 5 H)	364 366	C ₁₅ H ₁₃ BrN ₂ O ₂ S	49.50 (49.33)	3.66 (3.59)	7.64 (7.67)

Table 1. 2-Alkyl(aryl)-6-bromo-2,3,7,8-tetrahydro-3-thioxo-6,8a-ethanoimidazo[1,5-a]pyridine-1,5(6H,8aH)-diones (3)

^a ¹³C N.m.r. (CDCl₃-TMS): δ = 19.2(CH₂), 22.9(CH₂), 24.6(C), 36.5(CH₂), 50.2(CH₂), 52.4(C), 54.1(CH₂), 134.5, 135, 2 136.0, 142.9 (Ph), 182.1 (C=O), 185.5 (C=O), and 191.9 (C=S).

Table	2.	2-Alky	yl(ary	l)aminot	thiazole-5-	spirocy	clopro	pan-4(5 <i>H</i>	<i>I</i>)-ones (5)

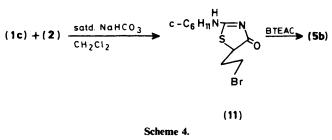
								Found (%) (Required)		
(5)	M.p. (°C)	Yield (%)	$v_{max}(KBr)$ (cm ⁻¹)	δ_{H} in CDCl ₃	$m/z (M^+)$	Formula	Ċ	Н	N	
a	140—141	15	3 210 (NH) 1 695 (C=O)	0.95—1.91 (m, $CH_2 \times 4$ and CH_3 , 11 H), 3.92 (t, CH_2N , 2 H)	198	$C_9H_{14}N_2OS$	54.71 (54.42)	7.19 (7.12)	14.36 (14.13)	
Ь	192—193	29	3 160 (NH) 1 690 (C=O)	1.13–1.87 (m, $CH_2 \times 7$, 14 H) 3.20 (m, CH , 1 H), 10.77 (br, NH , 1 H)	224	$C_{11}H_{16}N_2OS$	59.04 (58.90)	7.23 (7.19)	12.26 (12.49)	
c	99—100	18	3 260 (NH) 1 700 (C=O)	1.23 (m, CH ₂ , 2 H), 1.67 (m, CH ₂ , 2 H) 4.93 (s, CH ₂ N, 2 H), 7.33 (m, Ph, 5 H), 7.83 (br, NH, 1 H)		$C_{12}H_{12}N_2OS$	62.26 (62.04)	5.06 (5.21)	12.22 (12.06)	
d	179—180	22	3 200 (NH) 1 695 (C=O)	1.29 (m, CH ₂ , 2 H), 1.60 (m, CH ₂ , 2 H), 7.20–7.43 (m, Ph, 5 H), 7.90 (br, NH, 1 H)		$C_{11}H_{10}N_2OS$	60.56 (60.53)	4.80 (4.62)	12.62 (12.83)	
e	147—148	40	3 300 (NH) 1 680 (C=O)	1.42 (m, CH ₂ , 2 H), 1.80 (m, CH ₂ , 2 H), 7.45—7.43 (m, Naph, 7 H), 7.90 (br, NH, 1 H)		C ₁₅ H ₁₂ N ₂ OS	66.95 (67.14)	4.31 (4.51)	10.31 (10.44)	



(5) Scheme 3. (10)

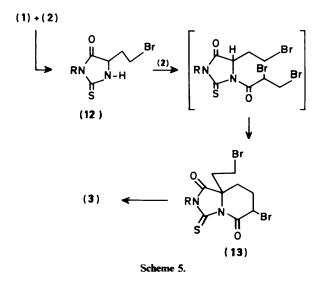
product supports structure (5), the conjugated imino group shifting the carbonyl absorption to lower frequency.

When the reaction of the thiourea (1) with the chloride (2) was performed in saturated NaHCO₃-CH₂Cl₂, 5-bromoethyl-thiazol-4(5*H*)-one (11) was afforded in 64-79% yields.



For the compound (5), the tautomer (10) was also considered. We have recently reported that the i.r. spectra of 2-iminothiazolidin-4-ones similar to (10) showed carbonyl absorptions at 1720-1735 cm^{-1.8} The carbonyl absorption observed in the Compound (11) was easily cyclized to the spiro compound (5) under phase-transfer conditions in fairly good yield.

The pathway of the reaction forming (3) from (1) and (2) is presumed to be as follows. Initially formed 4-bromoethyl-2thiohydantoin (12) reacts with additional compound (2) to afford the intermediate (13); this then undergoes intramolecular cyclization with elimination of hydrogen bromide to give (3). In fact, the intermediate (13) was isolated for the compound with $R=PhCH_2$.



Experimental

M.p.s were determined on a Yanagimoto micro-melting point apparatus and are uncorrected. I.r. spectra were recorded with a JASCO IRA-1 grating i.r. spectrometer. ¹H N.m.r. spectra were determined with a JEOL-60H spectrometer and ¹³C n.m.r. spectra were measured with a JEOL-FX-100 spectrometer using tetramethylsilane (TMS) as an internal standard. Mass spectra were measured with a JEOL-01 SG mass spectrometer.

 α,γ -Dibromobutyryl Chloride (2).— α,γ -Dibromobutyric acid was prepared from γ -butyrolactone and bromine according to Phillip's method,⁹ and then treated with SOCl₂ under reflux.

2-Alkyl(aryl)-6-bromo-2,3,7,8-tetrahydro-3-thioxo-6,8a-ethanoimidazo[1,5-a]pyridine-1,5(6H,8aH)-dione (3), the Corresponding Debromo Compound (4), and 2-Alkyl(aryl)aminothiazole-5-spirocyclopropan-4(5H)-one (5).-To a stirred mixture of thiourea (1) (5 mmol), CH₂Cl₂ (20 ml), and aqueous 5% NaOH (5 ml) was gradually added dibromobutyryl chloride (2) (2.64 g, 10 mmol) with ice-water cooling. Additional 5% NaOH (20 ml) and benzyltriethylammonium chloride (20 mg) were then added with stirring, and the latter was continued for 12 h at room temperature. The CH₂Cl₂ layer was separated, washed with water (15 ml \times 2), dried (Na₂SO₄), and evaporated. The residue was purified by a silica-gel column chromatography (CHCl₃-benzene, 8:2) [See Table 1 for compound (3) and Table 2 for compound (5)]: (4a) (0.15 g, 11%), m.p. 170-171 °C (Found: C, 58.85; H, 6.85; N, 10.55. C₁₃H₁₈N₂O₂S requires C, 58.62; H, 6.81; N, 10.52%); v_{max} (KBr) 1 703 (C=O), 1 684 (C=O), and 1 490 cm⁻¹ [NC(S)N]; $\delta_{H}(CDCl_{3})$ 0.95–2.00 (16 H, m, C_3H_7 , $CH_2 \times 4$, and CH), and 4.33 (2 H, t, CH_2N); m/z 266 (*M*⁺); (**4b**) (0.12 g, 8%), m.p. 228–229 °C (Found: C, 62.8; H, 4.8; N, 9.65. C₁₅H₁₄N₂O₂S requires C, 62.96; H, 4.93; N, 9.78%); v_{max} (KBr) 1729 (C=O), 1650 (C=O), and 1490 cm⁻¹ [NC(S)N]; δ_H(CDCl₃) 1.13—1.95 (2 H, m, CH₂), 2.47 (2 H, t, CH₂), 3.28 (1 H, t, CH), and 7.23–7.55 (5 H, m, Ph); m/z 286 $(M^{+}).$

Hydrolysis of 2-(1-Naphthylamino)thiazole-5-spirocyclopropan-4(5H)-one (5e).—A solution of compound (5e) (0.27 g, 12 mmol) in 15% NaOH (5 ml) and EtOH (5 ml) was refluxed for 4 h after which it was evaporated under reduced pressure; the alkaline solution was then extracted with CH_2Cl_2 (8 ml \times 2). The extract was washed with water (8 ml \times 2), dried (Na₂SO₄), and evaporated to afford 1-naphthylamine. The alkaline solution was acidified with 6M-HCl. The acidic solution was extracted with Et₂O (10 ml \times 2), and the extract was dried (Na₂SO₄) and evaporated to yield a residue which was recrystallized from EtOH to give bis(1-carboxycyclopropyl) disulphide (0.043 g, 36%), m.p. 194195 °C (Found: C, 40.85; H, 4.2. C₈H₁₀O₄ S requires C, 41.01; H, 4.30%); v_{max} (KBr) 1 700 cm⁻¹ (C=O); *m/z* 234 (*M*⁺).

2-Cyclohexylamino-5-bromoethylthiazol-4(5H)-one (11).—To a stirred solution of the thiourea (1c) (5 mmol) in saturated NaHCO₃ (8 ml)–CH₂Cl₂ (15 ml) was gradually added compound (2) (1.32 g, 5 mmol) (8 ml) with ice-water cooling. The reaction mixture was stirred for 8 h at room temperature after which the CH₂Cl₂ layer was separated, washed with water (10 ml × 2), dried (Na₂SO₄). Evaporation yielded a residue which was recrystallized from EtOH to give compound (11) (1.2 g, 79%), m.p. 171–172 °C (Found: C, 43.65; H, 5.55; N, 9.3. C₁₁H₁₇BrN₂O requires C, 43.28; H, 5.62; N, 9.18%); v_{max} (KBr) 3 160 (NH) and 1 700 cm⁻¹ (C=O); $\delta_{\rm H}$ (CDCl₃) 1.20–2.18 (13 H, m,c-C₆H₁₁ and CH₂), 3.57 (2 H, t, CH₂Br), 4.27 (1 H, m, CH), and 10.63 (1 H, br, NH); m/z 304, 306 (M⁺).

Conversion of 5-Bromoethyl-2-cyclohexylaminothiażol-4(5H)one (11) into 2-Cyclohexylaminothiazole-5-spirocyclopropan-4(5H)-one (5b).—A solution of compound (11) (2 mmol) in 5% NaOH (5 ml)—CH₂Cl₂ (10 ml) was stirred for 8 h in the presence of benzyltriethylammonium chloride (10 mg). The solution was worked up by the same method as described above.

Isolation of 2-Benzyl-6-bromo-8a-bromoethyl-2,3,7,8-tetrahydro-2-thioxoimidazo[1,5-a]pyridine-1,5(6H,8aH)-dione (13; $\mathbf{R} = PhCH_2$).—Treatment of compound (1d) with compound (2) in 5% NaOH-CH₂Cl₂ in the presence of benzyltriethylammonium chloride in the same method as described above afforded compound (13; $\mathbf{R} = PhCH_2$) after silica-gel column chromatography and recrystallization from EtOH, m.p. 212— 214 °C (Found: C, 41.45; H, 3.5; N, 5.95. C₁₆H₁₆Br₂N₂O₂ requires C, 41.76; H, 3.50; N, 6.09%); v_{max}(KBr) 1 730 (C=O), 1 650 (C=O), and 1 500 cm⁻¹ [NC(S)N]; m/z 459, 461 (M⁺).

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